ayny

Copper-Catalyzed Direct Trifluoromethylation of Propiolates: Construction of Trifluoromethylated Coumarins

Yuewen Li,[†] Yuan Lu,*^{*} Guanyinsheng Qiu,[§] and Qiuping Ding*^{*}

† Key Laboratory of Functio[nal](#page-2-0) Small Organic Molecules, Ministry of Education an[d C](#page-2-0)ollege of Chemistry & Chemical Engineering, Jiangxi Normal University, Nanchang, 330022, Jiangxi, China

‡ Obstetrics and Gynecology Hospital, Fudan University, 419 Fangxie Road, Shanghai 200011, China

§ College of Biological, Chemical Sciences and Engineering, Jiaxing University, Jiaxing 314001, Zhejiang, China

S Supporting Information

[AB](#page-2-0)STRACT: [A novel copp](#page-2-0)er-catalyzed direct trifluoromethylation of internal alkynes was developed, obtaining a series of trifluoromethylated coumarins in good yields. The cyclization was proposed to proceed via a radical mechanism under copper-catalyzed conditions with good functional group tolerance.

Cu(OAc)₂ (10 mol %) $K₂CO₃$ (2.0 equiv) CH₃CN, 60 °C 17 examples Togni's reagent

Coumarin and its derivatives rank among the most
important heterocycles in many naturally occurring
heterocyclic products and pharmaceutically active malagulas heterocyclic products and pharmaceutically active molecules. They have extensive applications as anticancer, antibacterial, anti-inflammatory, anti-HIV, antipsoriasis, anticoagulant, [or](#page-2-0) enzymatic inhibitors in medicinal chemistry, λ or as fluorescent probes and in illumination light source material sciences.³ Since it is generally accepted that smuggling a trifl[u](#page-2-0)oromethyl group (CF_3) into a potentially useful organic [m](#page-2-0)olecule could improve its physical and chemical as well as biological evaluations,⁴ some attention has been given to the trifluoromethyl substituted coumarins as anticancer agents, fluorescent markers, [o](#page-2-0)ptical molecular sensors, and polymer wavelength-tuning agents.⁵ Nonetheless, such applications are restricted to the easily available 4-trifluoromethyl substituted coumarins, which wer[e](#page-2-0) prepared by Lewis acid catalyzed condensation reactions from phenols.⁶ The limited successful but low yielding preparations of 3-trifluoromethyl substituted coumarins reported so far involve [th](#page-3-0)e reaction of coumarins with bis(trifluoroacetyl) peroxide and that of coumarin-3-carboxylic acids with sulfur tetrafluoride. 7 Additionally, the Bi group reported the copper-(I)-catalyzed regioselective C−H 3-trifluoromethylation of hymecromo[ne](#page-3-0) (an antispasmodic drug) using Togni's reagent.⁸ Therefore, the development of a general and practical protocol to afford 3-trifluoromethyl substituted coumarins fro[m](#page-3-0) accessible starting materials under mild reaction conditions remains highly desirable and equally challenging.

It has been of great synthetic interest to develop new methods for highly efficient and selective incorporation of the $CF₃$ group into diverse skeletal structures.⁹ Recently, many efforts have been focused toward the development of efficient means to access C−CF₃ bonds v[i](#page-3-0)a direct trifluoromethylation of alkenes using various CF_3 reagents.^{8,10,11} Additionally, regioselective difunctionalization of alkenes also provided a straightforward access to the $\text{C}_{(\text{sp}^3)}$ – CF_3 b[ond](#page-3-0).^{[12](#page-3-0)} Furthermore, the $CF₃$ group could be incorporated into many carbocycles and heterocycles using $TMSCF₃$ or Togni's reagent, such as oxindole, indoline, and isoquinoline-1,3-dione derivatives.¹³ Meanwhile, the 2-trifluoromethylated indole and 6-(trifluoromethyl)phenanthridine derivatives have been synthesiz[ed](#page-3-0) through oxidative cyclization of corresponding isocyanide derivatives with CF_3 reagents under metal-free conditions.¹⁴ However, to the best of our knowledge, only very few direct trifluoromethylations of alkynes have been reported.^{11a,12b[,15](#page-3-0)} Direct construction of $C_{\text{(sp}^2)}$ – CF_3 bonds through difunctionalization (oxy-trifluoromethylation) of terminal alky[nes has](#page-3-0) been developed by the groups of Szabó^{12b} [Scheme¹, eq 1], Sodeoka^{11a} and $\text{Cho}^{15\tilde{c}}$ [Scheme 1, eq 2] independently. During preparation of this Letter, three [oth](#page-3-0)er groups [re](#page-1-0)ported the dire[ct tr](#page-3-0)ifluorome[thyl](#page-3-0)ation of int[er](#page-1-0)nal alkynes.^{15d–f} Liu and Tan and co-workers described the transformation of propargylic alcohol[s](#page-3-0) into α -CF₃ enones using T[ogn](#page-3-0)i's reagent [Scheme 1, eq 3].^{15d} The Liang group reported the direct conversion of homopropargylic alcohols into CF_3 -containing 3butenal or 3-buten-[1-o](#page-3-0)ne derivatives in a regioselective manner [Scheme [1,](#page-1-0) eq 4].^{15e} The group of Ding and Hou developed the CuBr-catalyzed domino cyclization−trifluoromethylation of homopro[pa](#page-1-0)rgyl a[min](#page-3-0)es using Umemoto's reagent [Scheme 1, eq 5^{15f} Based on our recent efforts on the synthesis of trifluoromethylthiolated [he](#page-1-0)terocycles, 16 we herein present the direct [tr](#page-3-0)ifluoromethylation of propiolates 1 to obtain trifluoromethylated coumarins 3 [us](#page-3-0)ing Togni's reagents [Scheme 1, eq 6].

To verify the feasibility of the projected direct trifluoromethyla[tio](#page-1-0)n of propiolates illustrated in Scheme 1, we started to explore the practicability of this transformation. Initially, the substrate p-tolyl 3-phenylpropiolate 1a was sel[ec](#page-1-0)ted as the model. It has been reported that radicals can be generated from TMSCF₃ with $PhI(OAc)$ ₂ and the Togni reagent in the

Received: July 3, 2014 Published: August 1, 2014

Scheme 1. Direct Trifluoromethylation of Alkynes Table 1. Optimization of the Reaction Conditions^a

presence of copper.13d Based on these previous results, the reaction of 1a with TMSCF₃ in the presence of $PhI(OAc)₂$ in CH₃CN at 90 °C [was](#page-3-0) performed (Table 1, entry 1). Only a trace amount of the desired product 3a was detected. Pleasingly, the targeted product 3a was obtained in 15% yield when 1a reacted with Togni's reagent I in the presence of CuI. Then various catalysts were surveyed, revealing that $Cu(OAc)₂$ is best suited to running this reaction to provide 3a in 30% yield (Table 1, entries 3−8). To our delight, the yield was increased sharply when 2.0 equiv of K_2CO_3 were added in the reaction (Table 1, entry 9). Then we transferred our attention on other CF_3 reagents (Table 1, entries 10 and 11), but observed only poorer results. Changing the temperature of the reaction, we found the slightly effective temperature was at 60 °C with 55% yield (Table 1, entry 14). Gratifyingly, a satisfied yield (60%) was obtained when the quantity of Togni's reagent I was increased to 1.5 equiv. Unfortunately, the result could not be enhanced any more by switching the additive or solvent (see the Supporting Information).

With the optimized reaction conditions in hand, we started to [explore the substrate sc](#page-2-0)ope of this transformation. The results are shown in Scheme 2. In general, 3-arylpropiolate substrates 1 containing either electron-withdrawing or -donating groups reacted smoothly wi[th](#page-2-0) Togni's reagent I to give the corresponding trifluoromethylated coumarins in moderate to good yields. The effects of the substituent group $R¹$ were obvious. Results indicated that the substrates with electrondonating groups showed better reactivity (3c, 3f−g vs 3d−e). Additionally, phenyl 3-cyclopropylpropiolate 1h $(R¹ = cycle$ propyl) was also found to be suitable for the reaction under the standard conditions, providing the corresponding product in a relative low yield (3h). Subsequently, substrates (1i−k) bearing a methyl group $(R^2 = Me)$ in the *para-*, *meta-*, or *ortho-position* of the phenyl were subjected to the optimized reaction conditions. The results showed that the steric effect was also significant. Substrate 1i ($\mathbb{R}^2 = p$ -Me) was converted into the

	Ph $CF3$ reagent 1a 2 CF ₃	catalyst additive solvent, temp CF ₃	Ph 3a	CF ₃
	Togni reagent l		BF ₄ Umemoto reagent	
		Togni reagent II		
entry	$CF3$ reagent	catalyst	temp	yield $(\%)$
$\mathbf{1}$	$PhI(OAc)$, + TMSCF ₃		90	trace
2	Togni reagent I	CuI	90	15
3	Togni reagent I	CuBr	90	23
$\overline{4}$	Togni reagent I	CuOAc	90	17
5	Togni reagent I	[Cu(CN) ₄]PF ₆	90	nd
6	Togni reagent I	Cu ₂ O	90	28
7	Togni reagent I	CuBr ₂	90	24
8	Togni reagent I	Cu(OAc) ₂	90	30
9^b	Togni reagent I	$Cu(OAc)$ ₂	90	45
10^b	Togni reagent II	Cu(OAc) ₂	90	35
11^b	Umemoto reagent	$Cu(OAc)$ ₂	90	34
12^b	Togni reagent I	$Cu(OAc)$,	80	47
13^b	Togni reagent I	Cu(OAc) ₂	70	53
14^b	Togni reagent I	Cu(OAc) ₂	60	55
15^b	Togni reagent I	Cu(OAc) ₂	45	45
$16^{b,c}$	Togni reagent I	Cu(OAc) ₂	60	60
$17^{b,c,d}$	Togni reagent I	Cu(OAc),	60	58

a Unless otherwise noted, the reaction conditions are as follows: 1a (0.2 mmol), 2 (1.2 equiv, 0.24 mmol), catalyst (10 mol %), additive (2.0 equiv) , solvent (4 mL) , reaction time 12 h. ¹⁹F NMR yield of the isolated product. Isolated yield based on methyl 2-(phenylethynyl) benzoate $1a.$ b 2.0 equiv of K_2CO_3 were added. C Togni reagent I (1.5 equiv, 0.3 mmol). d_5 mol % catalyst $Cu(OAc)_2$ was used.

corresponding product 3i in good yield. While two isomers 3j and 3j′ were obtained in a ratio of 5:2 in a moderate total yield, and no desired product 3k was formed under the standard conditions. Furthermore, we also investigated the effects of substituent group R^2 . Different functional groups, including methyl, tertiary butyl, methoxyl, and halogen, were all compatible during the reaction process with good yields (3l− q). Painfully, no reaction took place when the activated alkyne was replaced by unactivated alkyne [1-methoxy-4-(3-phenoxyprop-1-yn-1-yl)benzene, 1r].

To probe into the possible reaction mechanism, 2,2,6,6 tetramethyl-1-piperidinyloxy (TEMPO), an effective radical scavenger, was added in the reaction under the standard conditions. As expected, a trace of the desired product (3a) was detected, and a distinctive compound 4 (¹⁹F NMR δ –56.06; see the Supporting Information) was observed instead. This result indicated that a CF_3 radical intermediate might be involved [in the reactions. Thu](#page-2-0)s, a plausible mechanism is proposed in Scheme 3. First, CF_3 radical A would be afforded under the proper conditions from Togni's reagent in the presence of copper([II\)](#page-2-0). The CF₃ radical A would be easily trapped by phenyl 3-phenylpropiolate 1, which would undergo a radical addition leading to imdoyl radical B. Subsequently, there are two possible ways for the formation of cyclohexadienyl cation E from the imdoyl radical B. In path a, imdoyl radical B undergoes radical addition to generate cyclohexadienyl radical C, which would be oxidized by

Scheme 3. Investigation of the Mechanism and the Proposed Mechanism

copper(III) to give cyclohexadienyl cation E. And in path b, the imdoyl radical B initially went through a single electron transfer (SET), followed by a Friedel−Crafts reaction also producing intermediate E. Finally, E underwent deprotonation to produce the targeted product 3.

In conclusion, we have reported a copper-catalyzed direct trifluoromethylation of activated alkynes, providing a rapid access to trifluoromethylated coumarins in good yields. Preliminary mechanistic studies indicate that the reaction proceeds through a CF_3 radical addition to activated alkynes, followed by sequential oxidation cyclization to target products. Further investigations on direct trifluoromethylation of internal alkynes for the synthesis of other heterocycles with privileged structures are underway in our laboratory.

■ ASSOCIATED CONTENT

S Supporting Information

Experimental procedures, Table S1, and product characterizations. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATION

Corresponding Authors

*E-mail: pipiluyuan@163.com.

*E-mail: dqpjxnu@gmail.com.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

Financial support from the National Natural Science Foundation of China (21262016), the Project of Jiangxi Youth Scientist (20122BCB23012), Jiangxi Educational Committee (GJJ12169), visiting programme of Jiangxi Normal University graduate, and Natural Science Foundation of Jiangxi Province of China (20133ACB20008, 20132BAB203006) is gratefully acknowledged.

■ REFERENCES

(1) (a) Naser-Hijazi, B.; Stolze, B.; Zanker, K. Second Proceedings of the International Society of Coumarin Investigators; Springer: Berlin, 1994. (b) Santana, L.; Uriarte, E.; Roleira, F.; Milhazes, N.; Borges, F. Curr. Med. Chem. 2004, 11, 3239. (c) Borges, F.; Roleira, F.; Milhazes, N.; Santana, L.; Uriarte, E. Curr. Med. Chem. 2005, 12, 887. (d) Trost, B. M.; Tost, F. J. Am. Chem. Soc. 1996, 118, 6305.

(2) (a) Peng, X.; Damu, G.; Zhou, C. Curr. Pharm. Des. 2013, 19, 3884. (b) RajeshwarRao, V.; Srimanth, K.; VijayaKumar, P. Indian J. Heterocyclic Chem. 2004, 14, 141−144.

(3) (a) Sokkalingam, P.; Lee, C.-H. J. Org. Chem. 2011, 76, 3820. (b) Xu, Q.; Ouyang, J.; Yang, Y.; Ito, T.; Kido, J. Appl. Phys. Lett. 2003, 83, 4695. (c) Lee, M.-T.; Yen, C.-K.; Yang, W.-P.; Chen, H.-H.; Liao, C.-H.; Tsai, C.-H.; Chen, C. Org. Lett. 2004, 6, 1241. (d) Zabradink, M. The Production and Application of Fluorescent Brightening Agent; John Wiley and Sons: New York, 1992.

(4) (a) Kirsch, P. Modern Fluoroorganic Chemistry: Synthesis, Reactivity, Applications; Wiley-VCH: Weinheim, 2004. (b) Ojima, I. Fluorine in Medicinal Chemistry and Chemical Biology; Wiley-Blackwell: Chichester, U.K., 2009. (c) Shimizu, M.; Hiyama, T. Angew. Chem., Int. Ed. 2005, 44, 214. (d) Müller, K.; Faeh, C.; Diederich, F. Science 2007, 317, 1881. (e) Schlosser, M. Angew. Chem., Int. Ed. 2006, 45, 5432. (f) Grate, J. W.; Frye, G. C. In *Sensors Update, Vol. 2*; Baltes, H., Göpel, W., Hesse, J., Eds.; Wiley-VCH: Weinheim, 1996; pp 10−20. (g) Bioorganic and Medicinal Chemistry of Fluorine; Begue, J.-P., Bonnet-Delpon, D., Eds.; Wiley: Hoboken, NJ, 2008. (h) Fluorine in Medicinal Chemistry and Chemical Biology; Ojima, I., Ed.; Wiley: Chichester, U.K., 2009. (i) Biffinger, J. C.; Kim, H. W.; DiMagno, S. G. ChemBioChem 2004, 5, 622. (j) Jeschke, P. ChemBioChem 2004, 5, 570.

(5) (a) Schill, H.; Nizamov, S.; Bottanelli, F.; Bierwagen, J.; Belov, V. N.; Hell, S. W. Chem.-Eur. J. 2013, 19, 16556. (b) Haishi, M.; Yamamoto, T.; Murata, M.; Ohtani, N. Journal of Physics: Conference Series 2008, 109, 012011. (c) Kubo, H.; Nagahori, H.; Tomigahara, Y.; Takahashi, J.; Harada, K.; Tanaka, A. PCT Int. Appl. WO 2009031706A120090312(2009). (d) Corrie, J.; Munasinghe, V.; Rettig, W. J. Heterocyclic Chem. 2000, 37, 1447. (e) Nad, S.; Pal, H. J. Phys. Chem. A 2000, 104, 673. (f) Makings, L.; Zlokamik, G. PCT Int. App1. WO 0035900 (1994).

(6) (a) Augustine, J. K.; Bombrun, A.; Ramappa, B.; Boodappa, C. Tetrahedron Lett. 2012, 53, 4422. (b) Rajanarendar, E.; Ramesh, P.; Raju, S. Indian J. Heterocyclic Chem. 2011, 20, 217. (c) Bose, D. S.; Rudradas, A. P.; Babu, M. H. Tetrahedron Lett. 2002, 43, 9195. (d) Nishiwaki, T.; Kikukawa, H. J. Heterocyclic Chem. 1994, 31, 889. (e) Bissel, E. R.; Larson, D. K.; Croudace, M. J. Chem. Eng. Data 1981, 26, 348. (f) Bissel, E. R.; Mitchell, A. R.; Smith, R. E. J. Org. Chem. 1980, 45, 2283.

(7) (a) Dmowski, W.; Piasecka-Maciejewska, K. Org. Prep. Proced. Int: The New Journal for Organic Synthesis 2002, 34, 514. (b) Matsui, M.; Shibata, K.; Muramatsu, H.; Sawada, H.; Nakayama, M. Synlett 1991, 113.

(8) Fang, Z.; Ning, Y.; Mi, P.; Liao, P.; Bi, X. Org. Lett. 2014, 16, 1522.

(9) For reviews for trifluoromethylation, see: (a) Schlosser, M. Angew. Chem., Int. Ed. 2006, 45, 5432. (b) Ma, J.-A.; Cahard, D. J. Fluorine Chem. 2007, 128, 975. (c) Ma, J.-A.; Cahard, D. Chem. Rev. 2008, 108, PR1. (d) Tomashenko, O. A.; Grushin, V. Chem. Rev. 2011, 111, 4475. (e) Studer, A. Angew. Chem., Int. Ed. 2012, 51, 8950. (f) Liu, X.; Xu, C.; Wang, M.; Liu, Q. Chem. Rev. 2014, DOI: 10.1021/ cr400473a. (g) Browne, D. Angew. Chem., Int. Ed. 2014, 53, 1482. (h) Liu, H.; Gu, Z.; Jiang, X. Adv. Synth. Catal. 2013, 355, 617. (i) Liu, X.; Wu, X. Synlett 2013, 24, 1882. (j) Egami, H.; Sodeoka, M. Angew. Chem., Int. Ed. 2014, 53, 10.1002/anie.201309260.

(10) (a) Parsons, A. T.; Buchwald, S. L. Angew. Chem., Int. Ed. 2011, 50, 9120. (b) Xu, J.; Fu, Y.; Luo, D.-F.; Jiang, Y.-Y.; Xiao, B.; Liu, Z.-J.; Gong, T.-J.; Liu, L. J. Am. Chem. Soc. 2011, 133, 15300. (c) Wang, X.; Ye, Y.; Zhang, S.; Feng, J.; Xu, Y.; Zhang, Y.; Wang, J. J. Am. Chem. Soc. 2011, 133, 16410. (d) Chu, L.; Qing, F.-L. Org. Lett. 2012, 14, 2016. (11) (a) Egami, H.; Shimizu, R.; Sodeoka, M. Tetrahedron Lett. 2012, 53, 5503. (b) Feng, C.; Loh, T.-P. Chem. Sci. 2012, 3, 3458. (c) Wang, X.-P.; Lin, J.-H.; Zhang, C.-P.; Xiao, J.-C.; Zheng, X. Beilstein J. Org. Chem. 2013, 9, 2635. (d) Feng, C.; Loh, T.-P. Angew. Chem., Int. Ed. 2013, 52, 12414. (e) Iqbal, N.; Choi, S.; Kim, E.; Cho, E. J. Org. Chem. 2012, 77, 11383.

(12) (a) Nguyen, J. D.; Tucker, J. W.; Konieczynska, M. D.; Stephenson, C. R. J. J. Am. Chem. Soc. 2011, 133, 4160. (b) Janson, P. G.; Ghoneim, I.; Ilchenko, N. O.; Szabó, K. J. Org. Lett. 2012, 14, 2882. (c) Li, Y.; Studer, A. Angew. Chem., Int. Ed. 2012, 51, 8221. (d) Yasu, Y.; Koike, T.; Akita, M. Angew. Chem., Int. Ed. 2012, 51, 9567. (e) Zhu, R.; Buchwald, S. L. J. Am. Chem. Soc. 2012, 134, 12462. (f) Mizuta, S.; Verhoog, S.; Engle, K. M.; Khotavivattana, T.; O'Duill, M.; Wheelhouse, K.; Rassias, G.; Médebielle, M.; Gouverneur, V. J. Am. Chem. Soc. 2013, 135, 2505. (g) Yasu, Y.; Koike, T.; Akita, M. Org. Lett. 2013, 15, 2136. (h) Egami, H.; Kawamura, S.; Miyazaki, A.; Sodeoka, M. Angew. Chem., Int. Ed. 2013, 52, 7841. (i) Chen, Z.-M.; Bai, W.; Wang, S.-H.; Yang, B.-M.; Tu, Y.-Q.; Zhang, F.-M. Angew. Chem., Int. Ed. 2013, 52, 9781. (j) Wang, F.; Qi, X.; Liang, Z.; Chen, P.; Liu, G. Angew. Chem., Int. Ed. 2014, 53, 1881. (k) He, Y.-T.; Li, L.- H.; Yang, Y.-F.; Zhou, Z.-Z.; Hua, H.-L.; Liu, X.-Y.; Liang, Y.-M. Org. Lett. 2014, 16, 270. (l) Carboni, A.; Dagousset, G.; Magnier, E.; Masson, G. Org. Lett. 2014, 16, 1240. (m) Xu, T.; Cheung, C. W.; Hu, X. Angew. Chem., Int. Ed. 2014, 53, 4910.

(13) (a) Mu, X.; Wu, T.; Wang, H.; Guo, Y.; Liu, G. J. Am. Chem. Soc. 2012, 134, 878. (b) Egami, H.; Shimizu, R.; Kawamura, S.; Sodeoka, M. Angew. Chem., Int. Ed. 2013, 52, 4000. (c) Xu, P.; Xie, J.; Xue, Q.; Pan, C.; Cheng, Y.; Zhu, C. Chem.—Eur. J. 2013, 19, 14039. (d) Li, L.; Deng, M.; Zheng, S.-C.; Xiong, Y.-P.; Tan, B.; Liu, X.-Y. Org. Lett. 2014, 16, 504.

(14) (a) Wang, Q.; Dong, X.; Xiao, T.; Zhou, L. Org. Lett. 2013, 15, 4846. (b) Cheng, Y.; Jiang, H.; Zhang, Y.; Yu, S. Org. Lett. 2013, 15, 5520. (c) Zhang, B.; Mü ck-Lichtenfeld, C.; Daniliuc, C. G.; Studer, A. Angew. Chem., Int. Ed. 2013, 52, 10792. (d) Zhang, B.; Studer, A. Org. Lett. 2014, 16, 1216.

(15) (a) Kitazume, T.; Ishikawa, N. J. Am. Chem. Soc. 1985, 107, 5186. (b) Takeyama, Y.; Ichinose, Y.; Oshima, K.; Utimoto, K.

Tetrahedron Lett. 1989, 30, 3159. (c) Iqbal, N.; Jung, J.; Park, S.; Cho, E. J. Angew. Chem., Int. Ed. 2014, 53, 539. (d) Xiong, Y.-P.; Wu, M.-Y.; Zhang, X.-Y.; Ma, C.-L.; Huang, L.; Zhao, L.-J.; Tan, B.; Liu, X.-Y. Org. Lett. 2014, 16, 1000. (e) Gao, P.; Shen, Y.-W.; Fang, R.; Hao, X.-H.; Qiu, Z.-H.; Yang, F.; Yan, X.-B.; Wang, Q.; Gong, X.-J.; Liu, X.-Y.; Liang, Y.-M. Angew. Chem., Int. Ed. 2014, 53, 7629. (f) Ge, G.-C; Huang, X.-J.; Ding, C.-H; Wan, S.-L.; Dai, L.-X.; Hou, X.-H. Chem. Commun. 2014, 50, 3048.

(16) (a) Xiao, Q.; Sheng, J.; Ding, Q.; Wu, J. Eur. J. Org. Chem. 2014, 217. (b) Li, Y.; Li, G.; Ding, Q. Eur. J. Org. Chem. 2014, DOI: 10.1002/ejoc.201402629R1.